Matter Unit Topic 2 Study Guide

Use this as a reference guide on what to study for your assessment on Topic 2. If you know all of these answers, you should be good to go for your assessment!

- 1. What is the movement for the molecules in each of the three main states of matter?
- 2. Describe if each of the three main states of matter has a definite volume and/or shape.
- 3. What is Boyle's Law and what type of relationship does its graph have?
- 4. What is Charles's Law and what type of relationship does its graph have?
- 5. How are Boyle's Law and Charles's Law similar and how are they different?
- 6. What is surface tension and how can it be useful?
- 7. What is viscosity and what does it mean if a substance has a high or low viscosity?
- 8. What is vaporization and how does it change the state of matter for a substance?
- 9. What is boiling point and how does it change the state of matter for a substance?
- 10. What is freezing point and how does it change the state of matter for a substance?
- 11. What is evaporation and how does it change the state of matter for a substance?
- 12. What is condensation and how does it change the state of matter for a substance?
- 13. What is sublimation and how does it change the state of matter for a substance?
- 14. Increasing the temperature (or thermal energy) of a substance does what to that substance? Decreasing the temperature (or thermal energy) does what instead?
- 15. Using the picture below, describe the movement of molecules for each state of matter.

